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Answer the questions below by circling the number or the correct response 1.

Which of the following is NOT matter? (1)

a chair (2) air ... Evan P. Silberstein,

2002 . 1) A student measures the mass

of an 8 cm<sup>3</sup> block of brown sugar to be

12.9 g. What is the density of the brown

sugar?

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Based on your understanding of covalent bonds, answer the questions below. 1. Draw electron dot diagrams for hydrogen and oxygen. 2. Draw electron dot diagrams showing the pairing of electrons to form water from hydrogen and oxygen. All outer shells should be

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complete. 3. Are the bonds in water polar or nonpolar. How do you know?  
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acids and bases are caused by the ions they form in water. Due to the presence of ions, aqueous solutions of both acids and bases are electrolytes. Acids and bases react with each other to form a salt and water. The reaction is a double replacemnt reaction known as neutralization.

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answers to the problems below using the  
gas laws. \_\_\_\_\_ 1. What is the new

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volume of a gas if 50 mL at 81.0 kPa has its pressure increased to 101.3 kPa?

(Temperature is constant.) \_\_\_\_\_ 2. 720

mL of H<sub>2</sub> gas at 0°C and 126.6 kPa is changed to S.T.P. What will be its new volume ? \_\_\_\_\_ 3. 440 mL of N<sub>2</sub>

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Elements , Compounds

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Naming Substituted Hydrocarbons A substituted hydrocarbon is a hydrocarbon with an element other than hydrogen attached somewhere along the hydrocarbon chain. It is named in a

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similar fashion to a hydrocarbon. This can be illustrated with alcohols as an example. The compounds pictured to the lower left are alcohols. They look like alkanes

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That can't be broken down. When put together chemically, They always form



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compounds. But thinking of the usual way That chemicals are found,

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